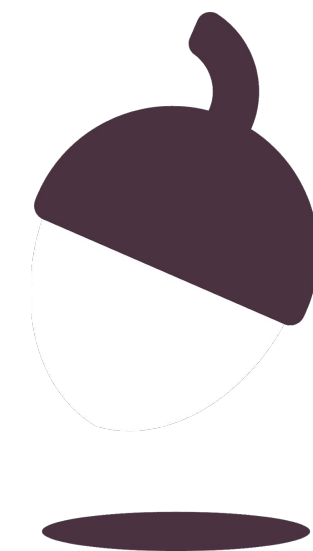


Combined Science - Biology - KS4
Cell Biology

Cell biology review 2

Miss Wong



OAK
NATIONAL
ACADEMY

Transport of materials



Transport into and out of cell

Transport

Osmosis

Water molecules only

Diffusion

All particles apart
from water molecules

Active transport

E.g glucose and
mineral ions



Diffusion

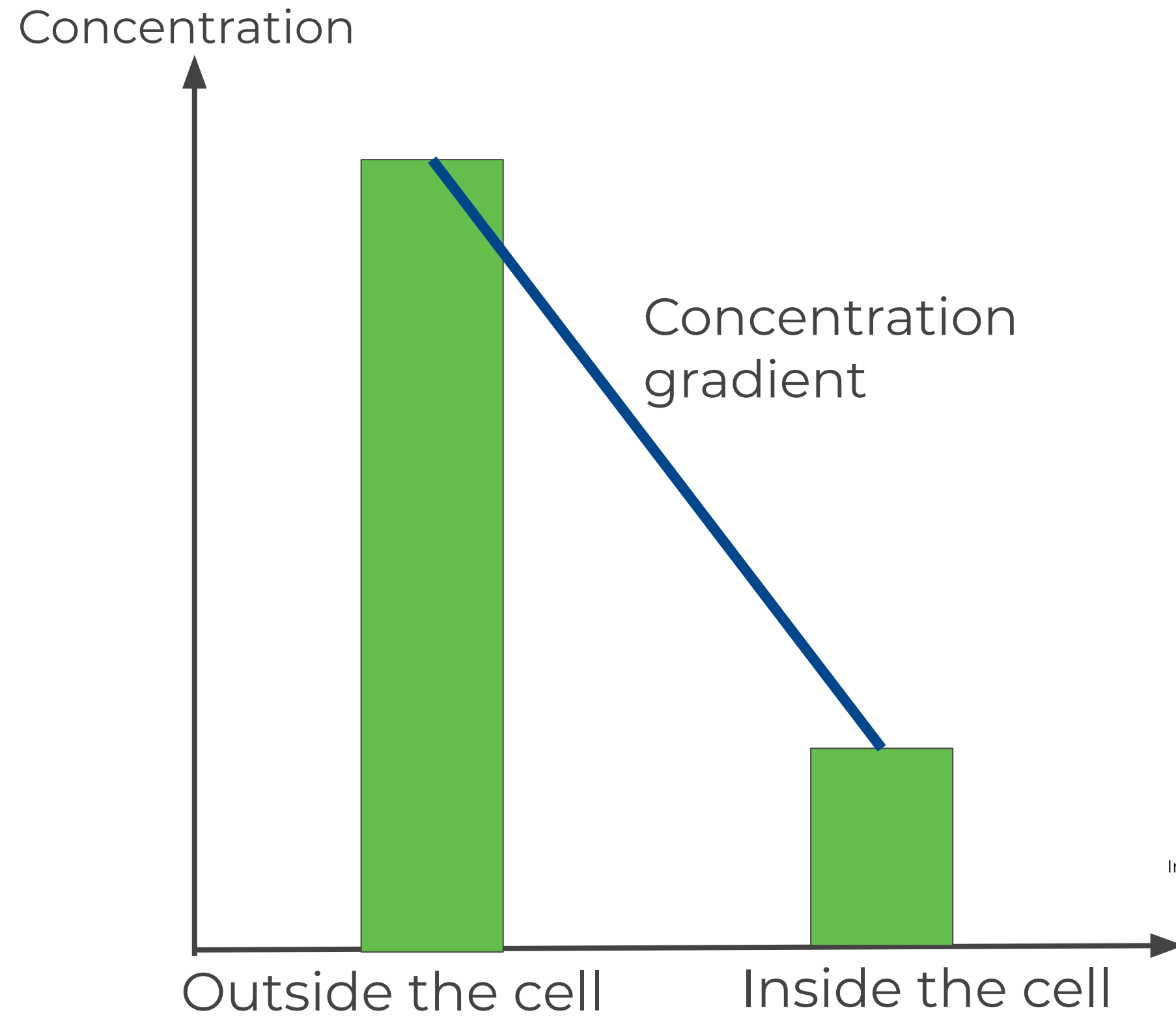


Image by Miss C. Wong, Oak National Academy



Diffusion

The concentration gradient will decrease as diffusion takes place.

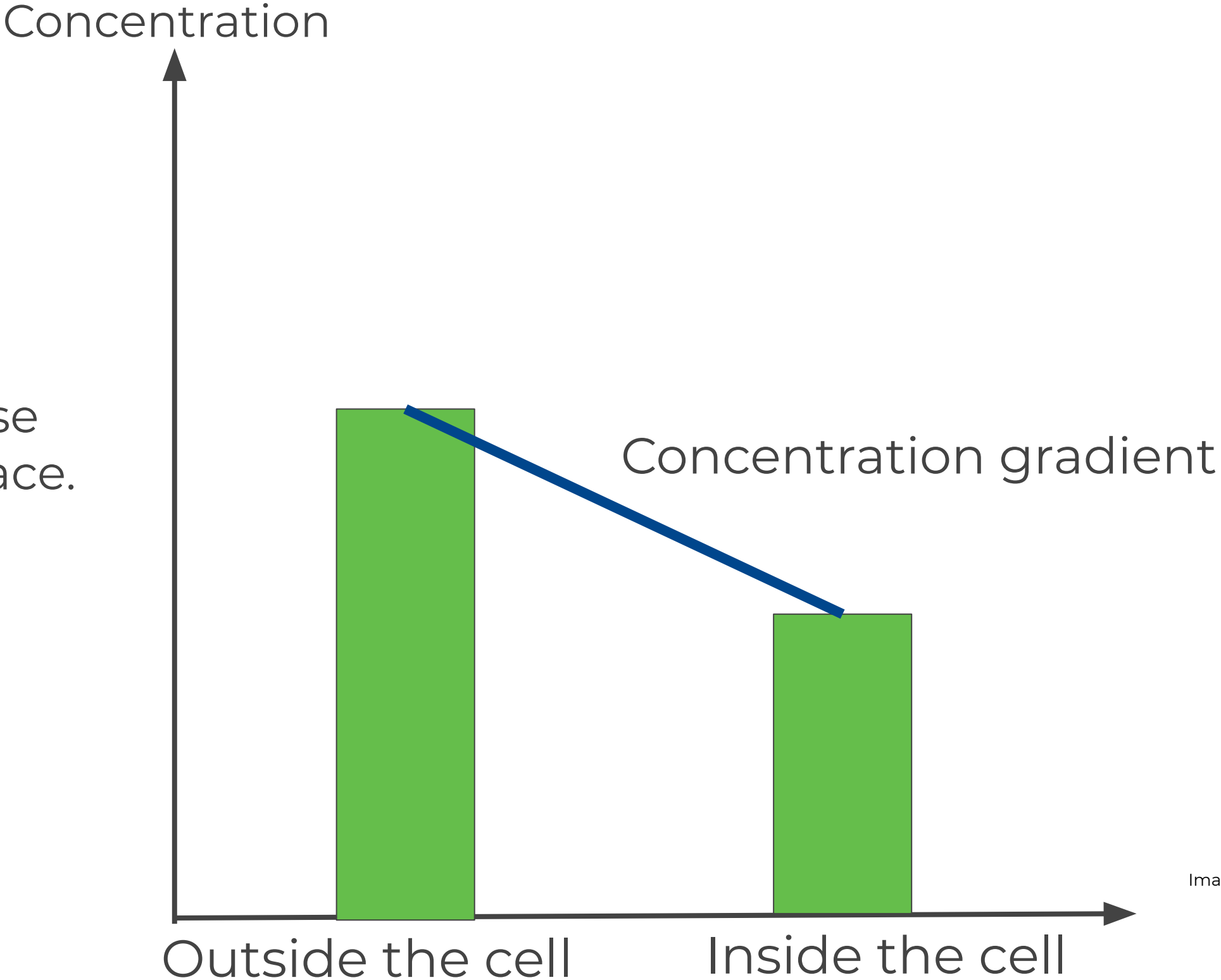
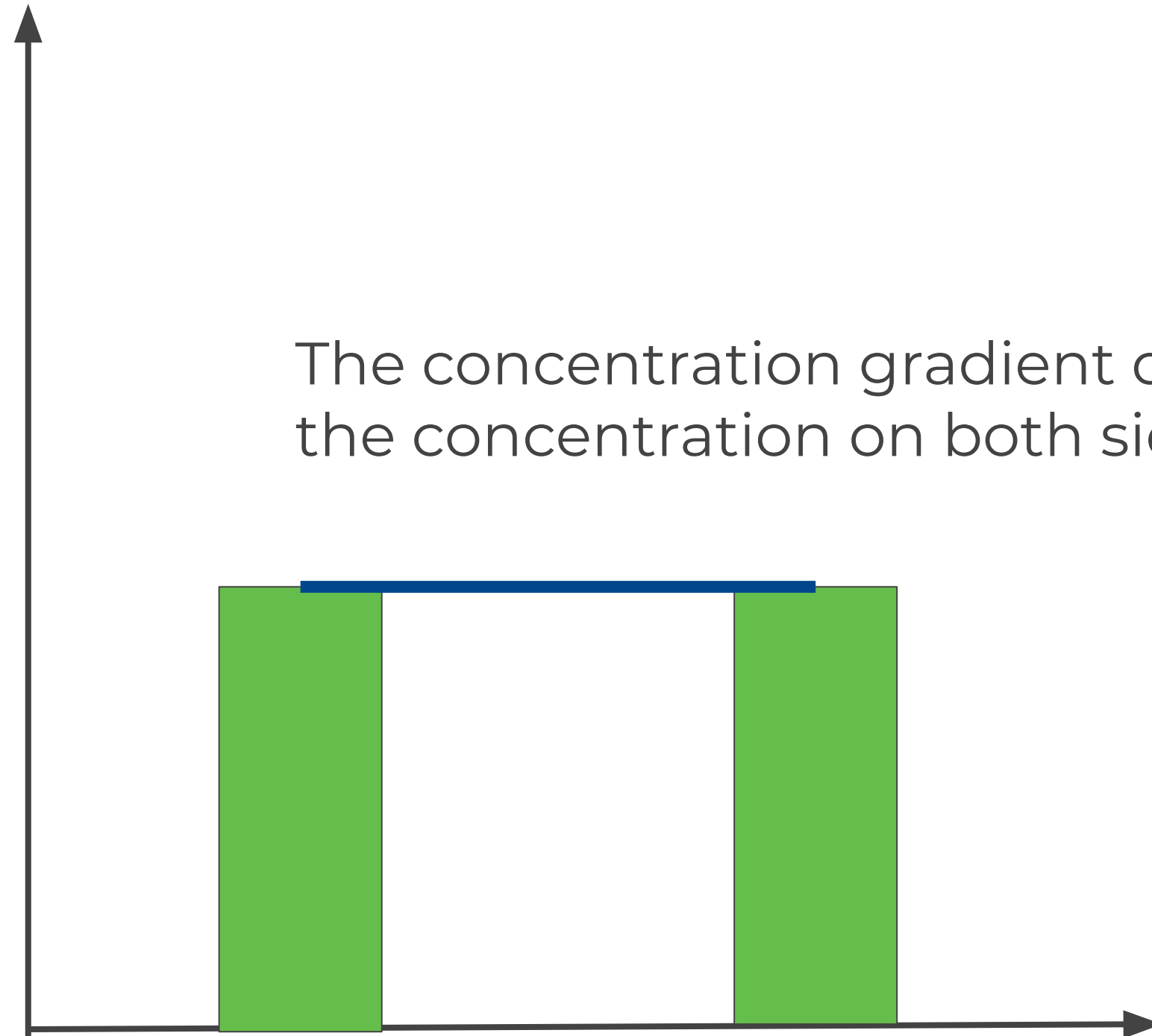


Image by Miss C. Wong, Oak National Academy



Diffusion

Concentration



The concentration gradient does not exist when the concentration on both sides are equal.

Outside the cell

Inside the cell

Image by Miss C. Wong, Oak National Academy



Active transport

Active transport is the movement of substances from a lower concentration to a higher concentration **across a semi-permeable** membrane **requiring energy** from respiration.

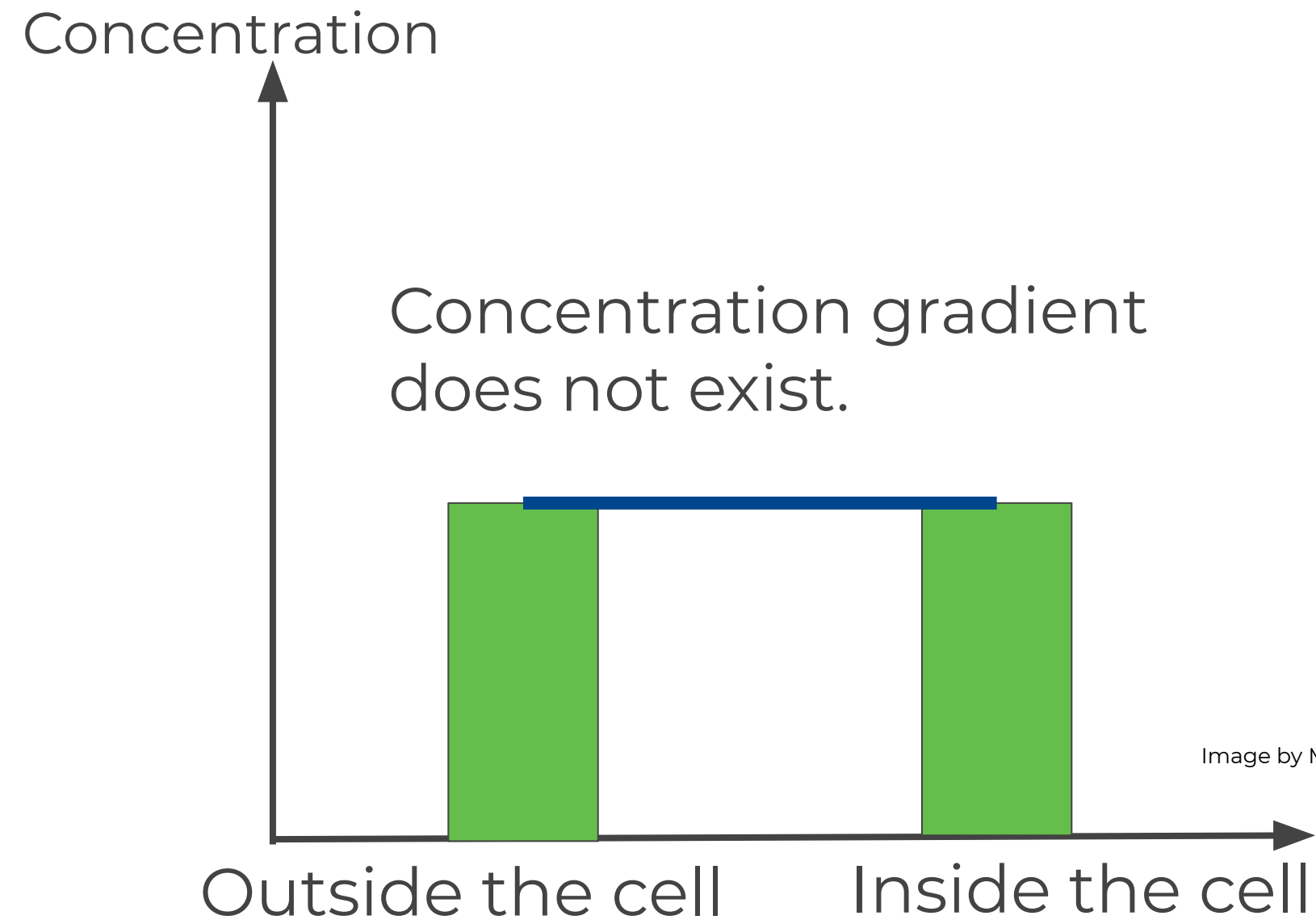
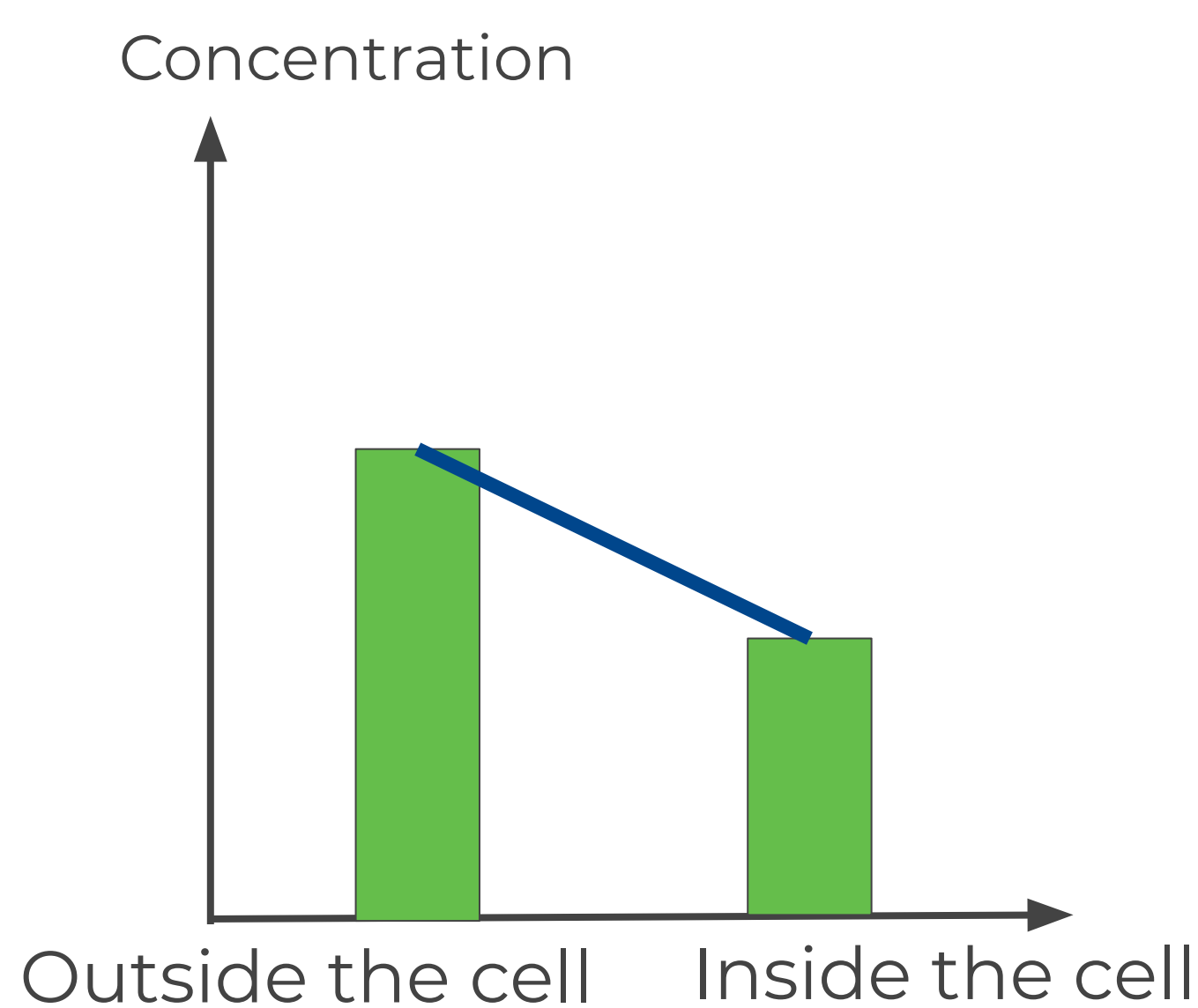
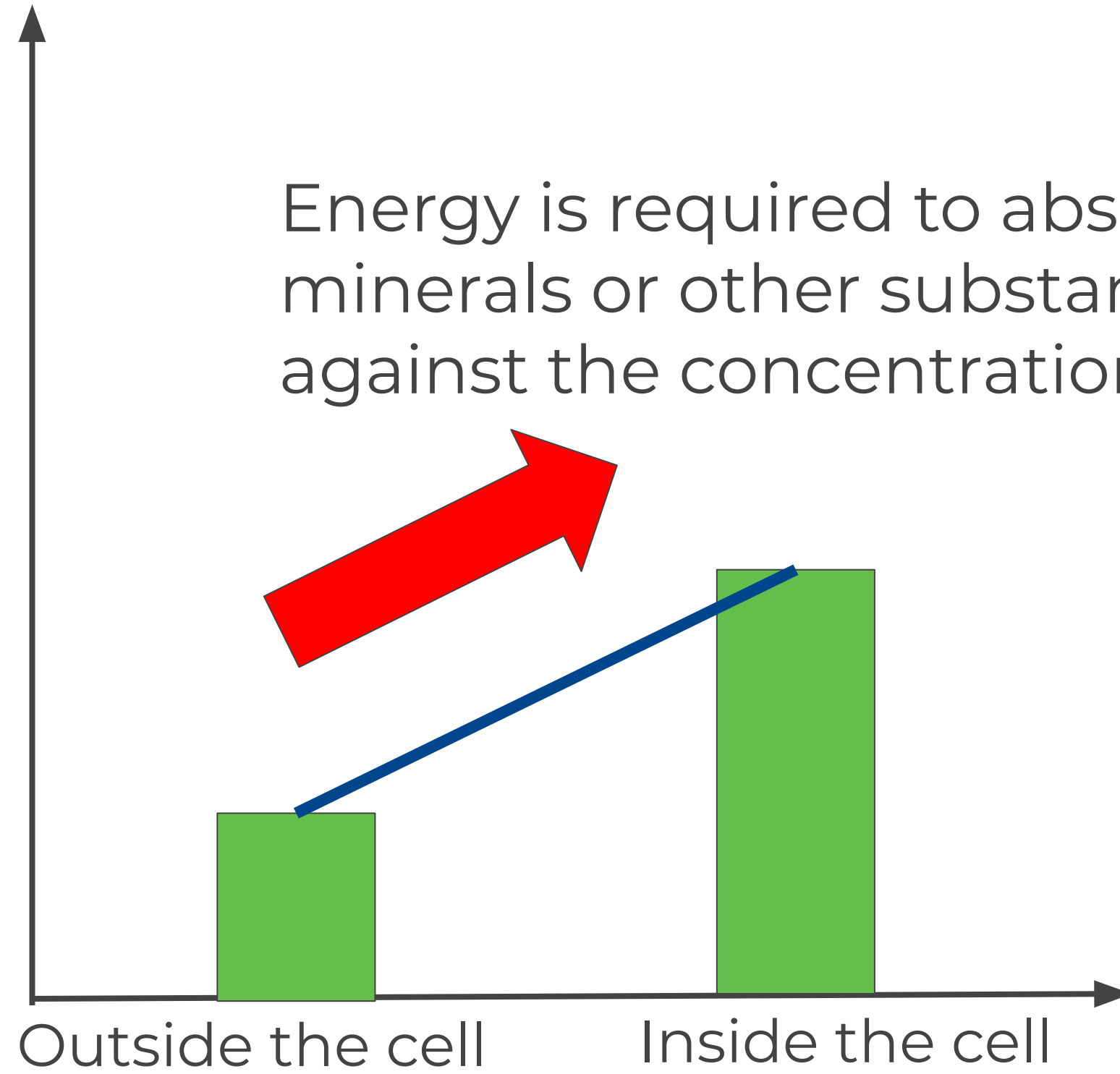


Image by Miss C. Wong, Oak National Academy



Active transport

Concentration



Energy is required to absorb minerals or other substances against the concentration gradient.

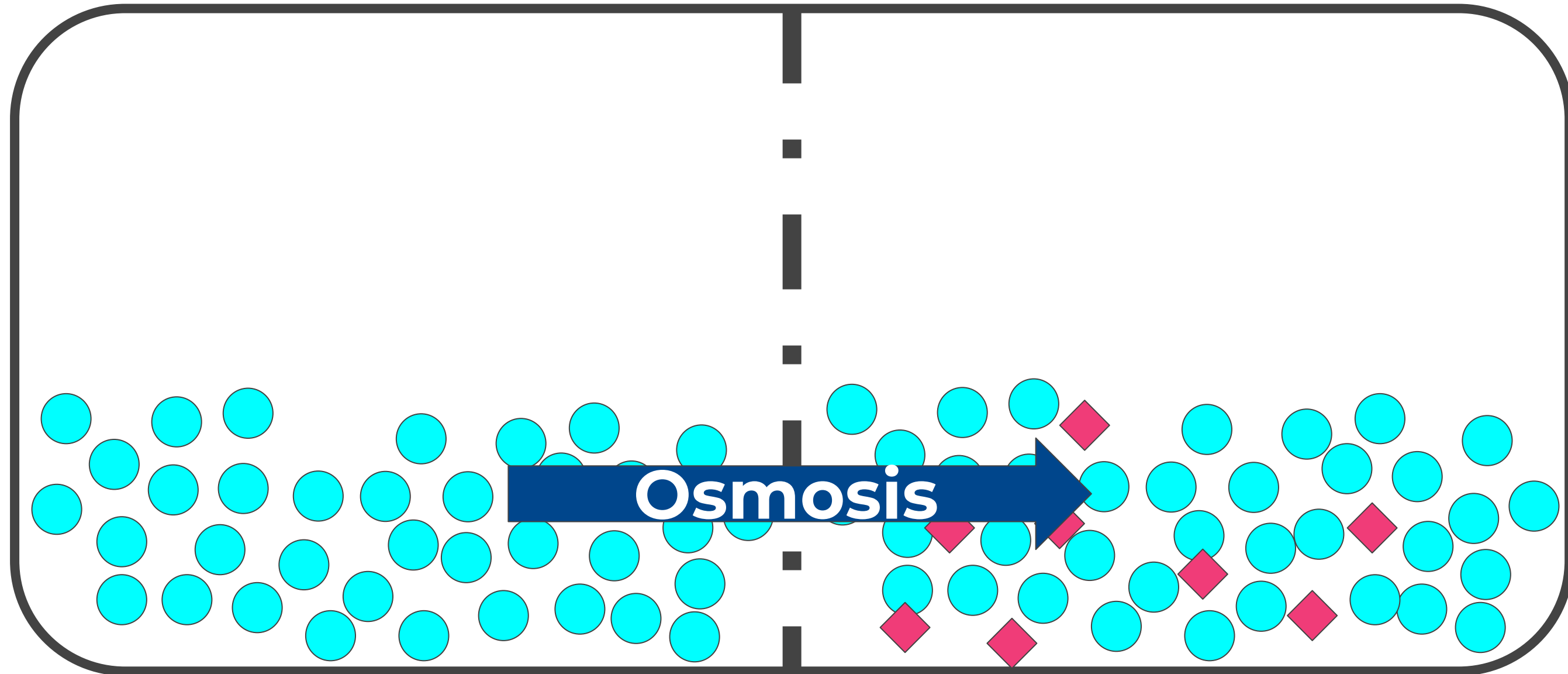
This process requires energy and proteins in the cell membrane.



Osmosis

The movement of water from a region of higher water concentration to a region with lower water concentration.

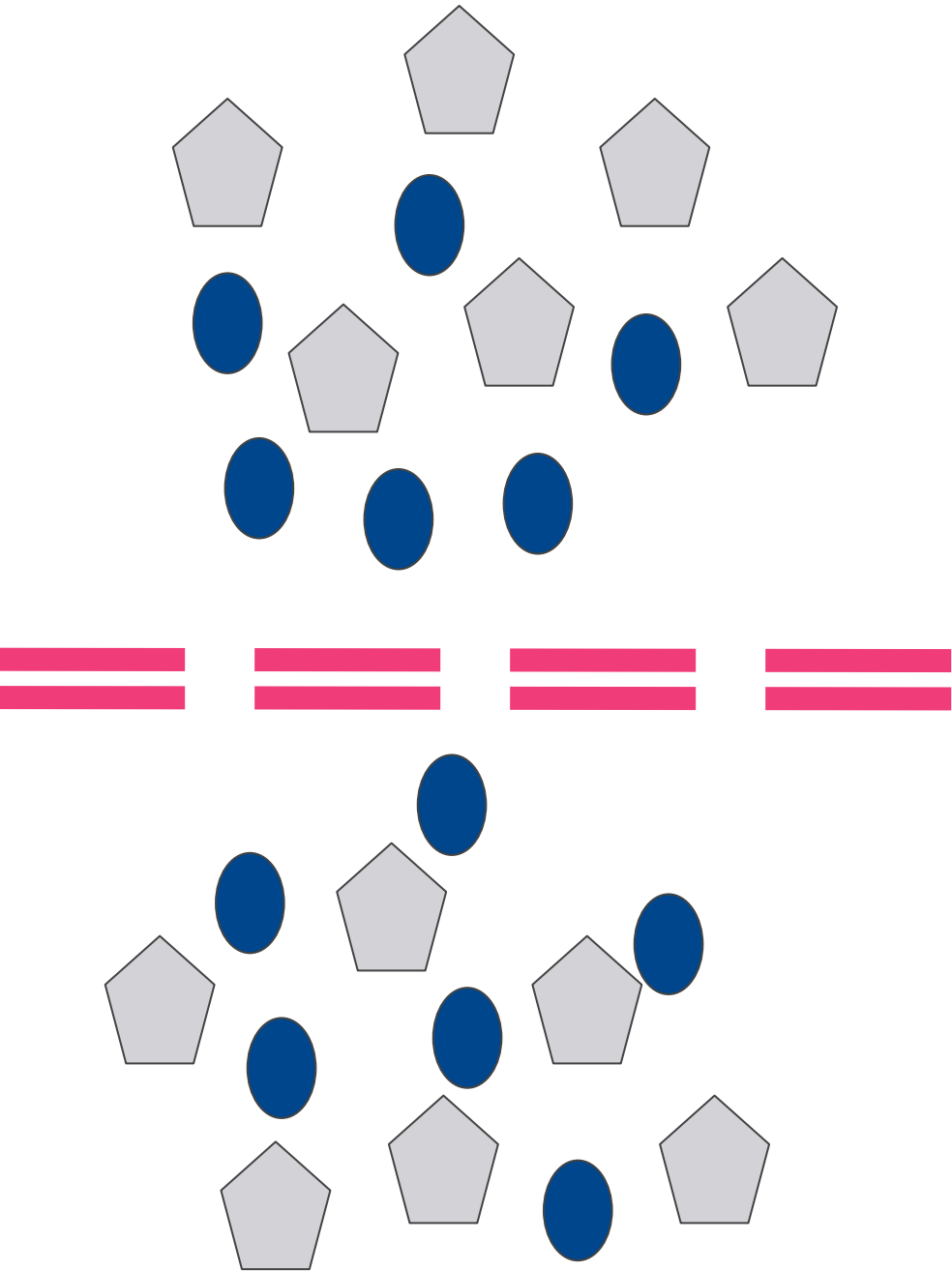
Image by Miss C. Wong, Oak National Academy



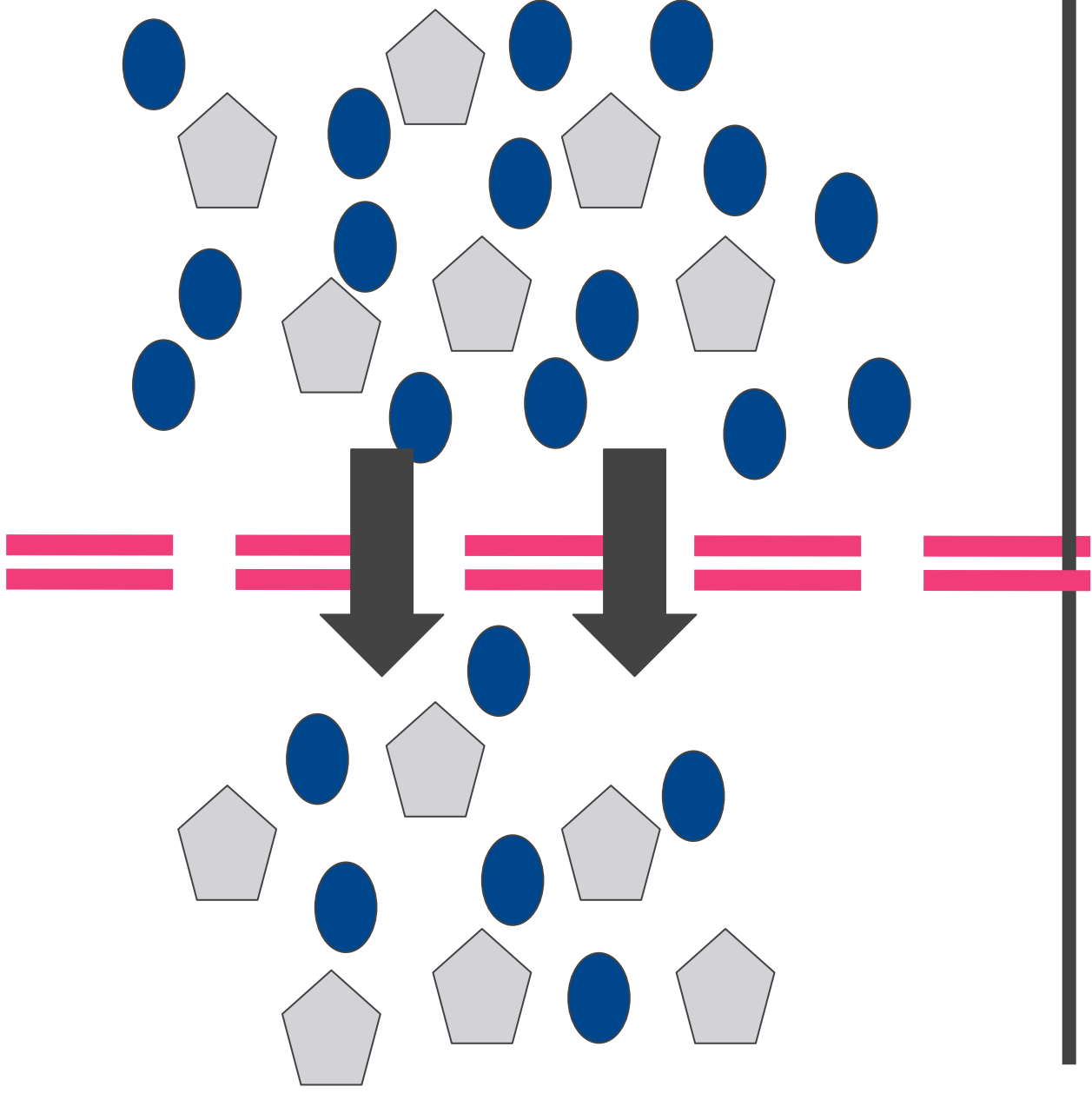
Hypotonic, hypertonic, isotonic solution

Image created by Miss C. Wong

Isotonic solution

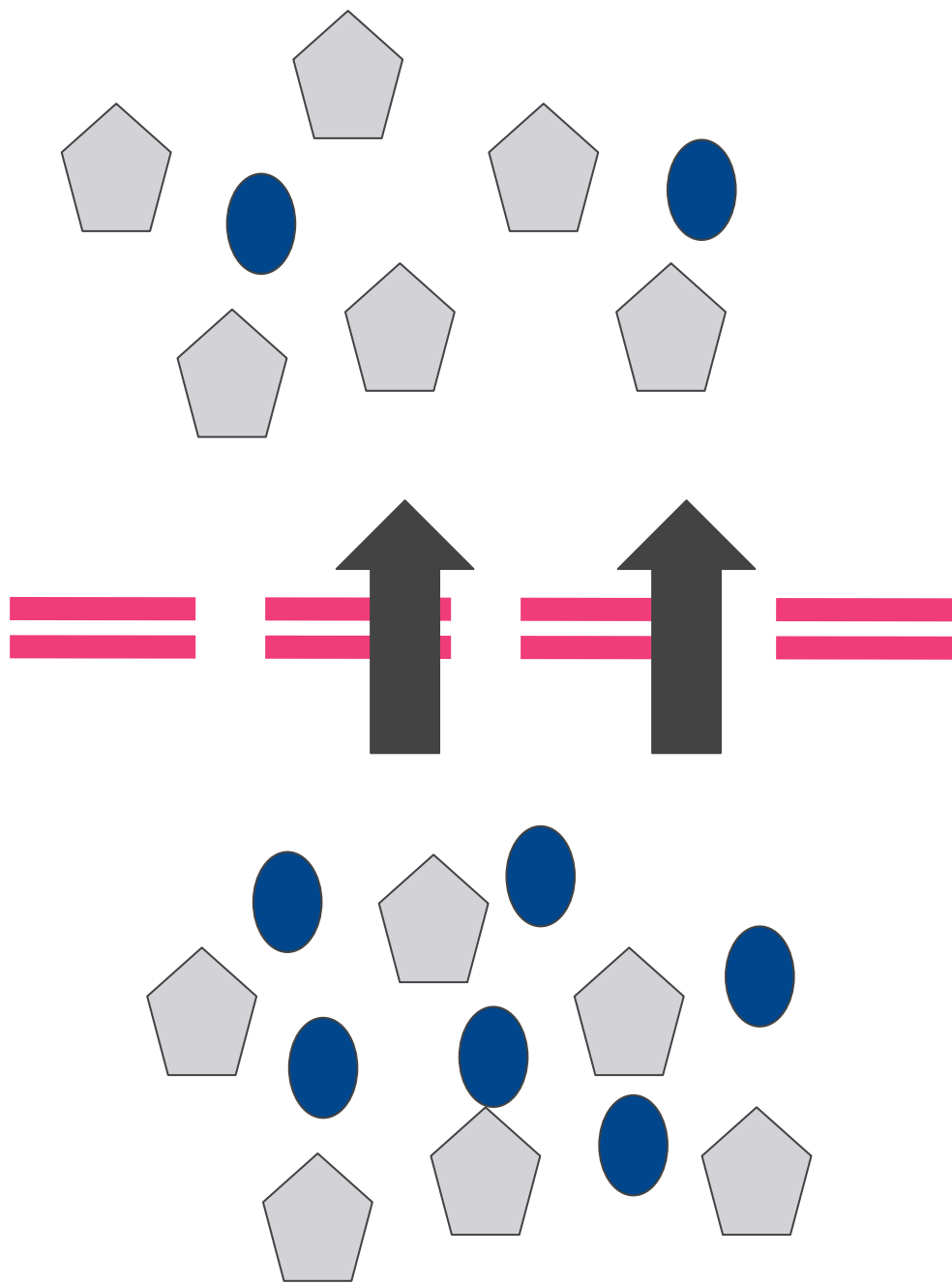


Hypotonic solution



Inside the cell

Hypertonic solution



Multiple choice quiz

Water goes into root hair cells by...

Osmosis

Diffusion

Active transport



Multiple choice quiz

Root hair cells absorb mineral ions against the concentration gradient by...

Osmosis

Diffusion

Active transport



Multiple choice quiz

Which two of the following require a cell membrane to occur?

Osmosis

Diffusion

Active transport



Multiple choice quiz

Which two of the following involves the movement of particles down the concentration gradient?

Osmosis

Diffusion

Active transport



Multiple choice quiz

Plasmolysis will happen when a plant cell is put into...

A hypotonic solution

An isotonic solution

A hypertonic solution



Factors affecting rate of transport



Four factors affecting rate of transport

Temperature

The higher the temperature, the higher the rate.

Difference in concentration

The bigger the difference, the higher the rate.

Surface area to volume ratio

The bigger the surface area to volume ratio, the higher the rate.

Diffusion distance

The shorter the diffusion distance, the higher the rate.



Examples of exchange surfaces

The examples are:

1. Root hair cells.
2. Alveoli
3. Fish lamellae
4. Villi

They all have:

1. Thin walls
2. Large surface area to volume ratio
3. Good blood supply (not in the root hair cell)

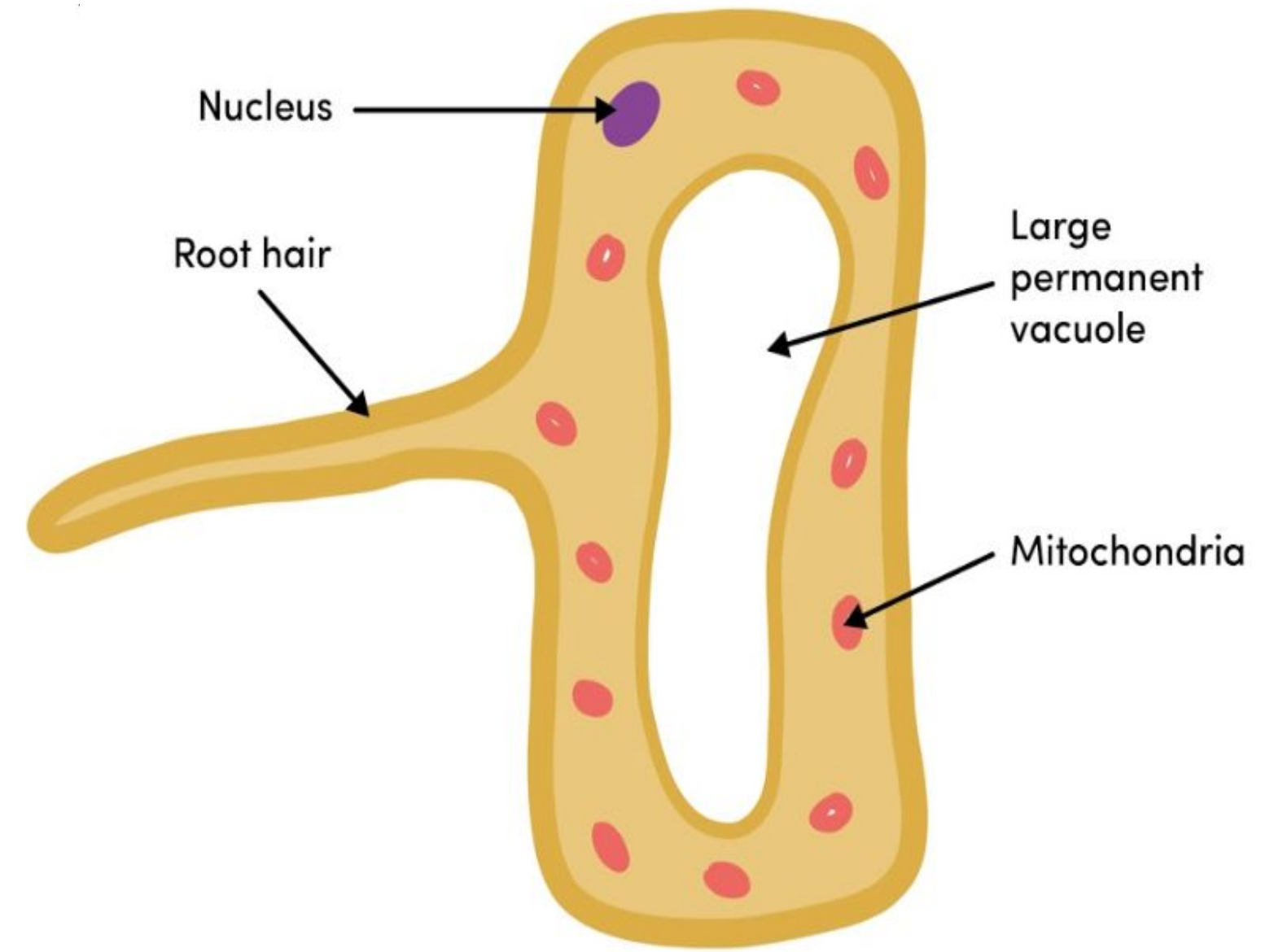


Image: Oak National Academy



Pause the video to complete your task

Quick concept check

Give three common adaptations of the human alveoli and the fish lamellae.

Resume once you're finished



Pause the video to complete your task

Answer

They both have thin walls, large surface area to volume ratio and they have good blood supply to maintain the concentration gradient.

Resume once you're finished



Osmosis required practical



Pause the video to complete your task

Quick reminder

1. Independent variable is the one you _____.
2. Dependent variable is the one you _____.
3. Control variable is the one you _____.

Resume once you're finished



Pause the video to complete your task

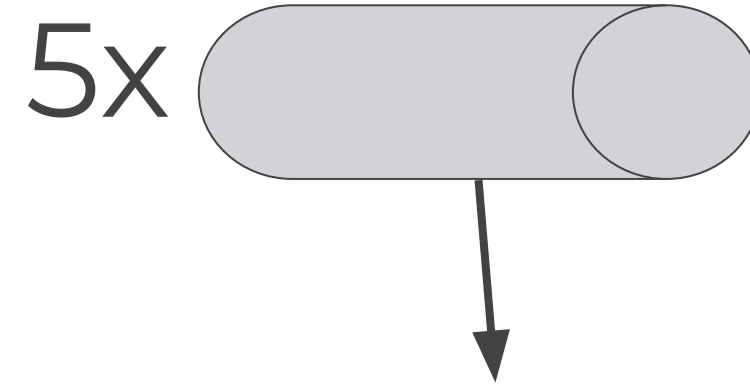
Answers

1. Independent variable is the one you change.
2. Dependent variable is the one you measure.
3. Control variable is the one you keep the same.

Resume once you're finished



What is the salt or sugar concentration inside a plant tissue?



Soaked into different concentration of salt or sugar solution

Increase in mass or length

No change

Decrease in mass or length

Same concentration

Water entered the cells by osmosis. Concentration of salt must be higher than the solution.

Water left the cells by osmosis. Concentration of salt must be lower than the solution.



Working scientifically as a process

What is the concentration of salt or sugar in the cells of a plant?



The concentration is 0.4 molar



Design experimental setup

Different concentrations of salt solution being used.

Time of plant tissue being soaked in the solution.



The change in mass and/or length.



Test and collect data



What is the concentration of salt or sugar?



Results table

Concentration of sugar solution/%	Starting length/cm	Final length/cm	Change in length/cm	Percentage change in length/ %
4	6.00	5.20		
3	6.00	5.70		
2	6.00	6.10		
1	6.00	6.40		
0	6.00	6.90		



Finding the percentage change

Steps 1: find the change

Step 2: apply

$$\text{Percentage change} = \frac{\text{change}}{\text{starting value}} \times 100$$

There is a piece of carrot. The carrot had a length of 6 cm before being put completely into 5% sugar solution. After five hours, the carrot was removed from the water, blotted dry and weighed. The mass of the carrot was 5.1 cm. Calculate the percentage change in mass.

$$\text{The change} = 5.1 - 6 = -0.9$$

$$\text{Percentage change} = -0.9 \div 6 \times 100 = -15.00\%$$



Results table

Concentration of sugar solution/%	Starting length/cm	Final length/cm	Change in length/cm	Percentage change in length/ %
4	6.00	5.20		
3	6.00	5.70		
2	6.00	6.10		
1	6.00	6.40		
0	6.00	6.90		

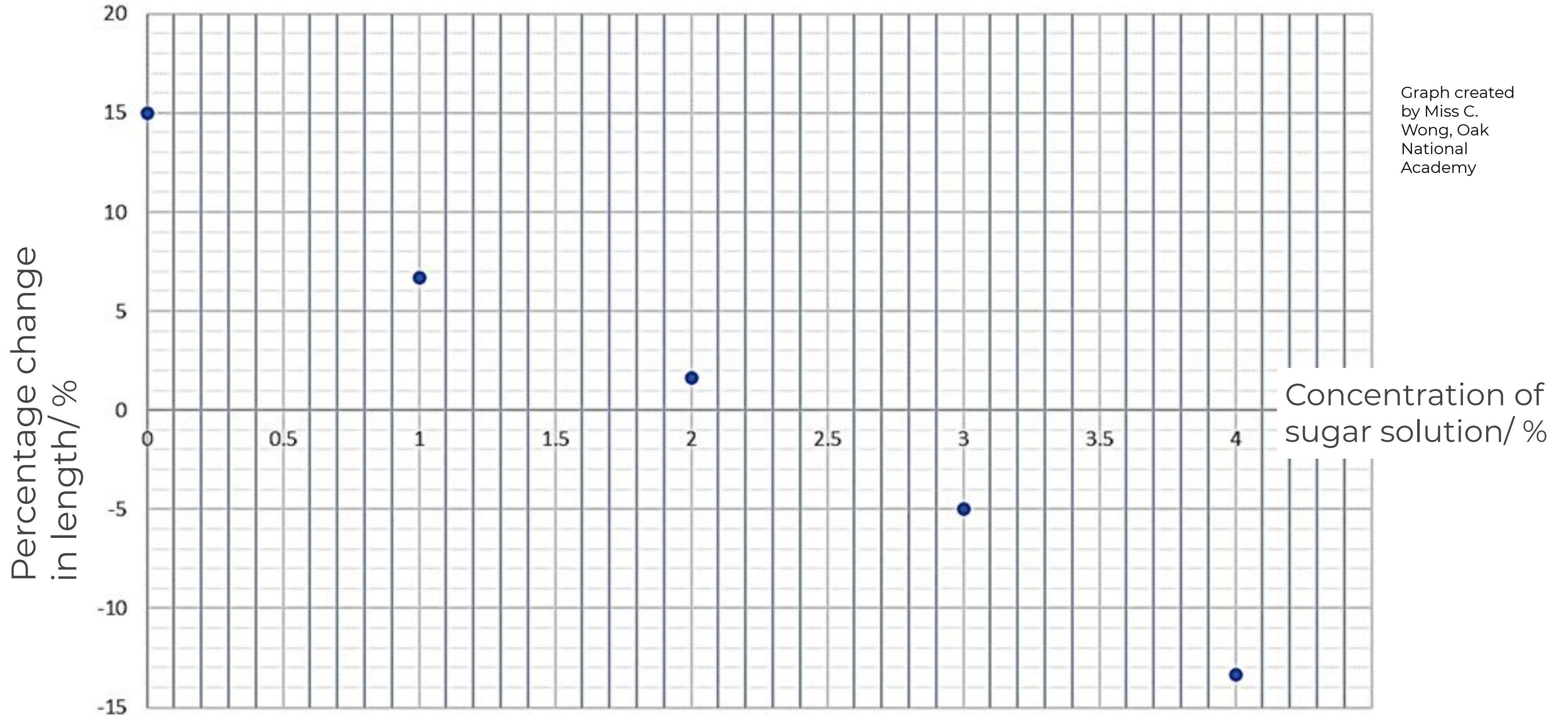


Results table

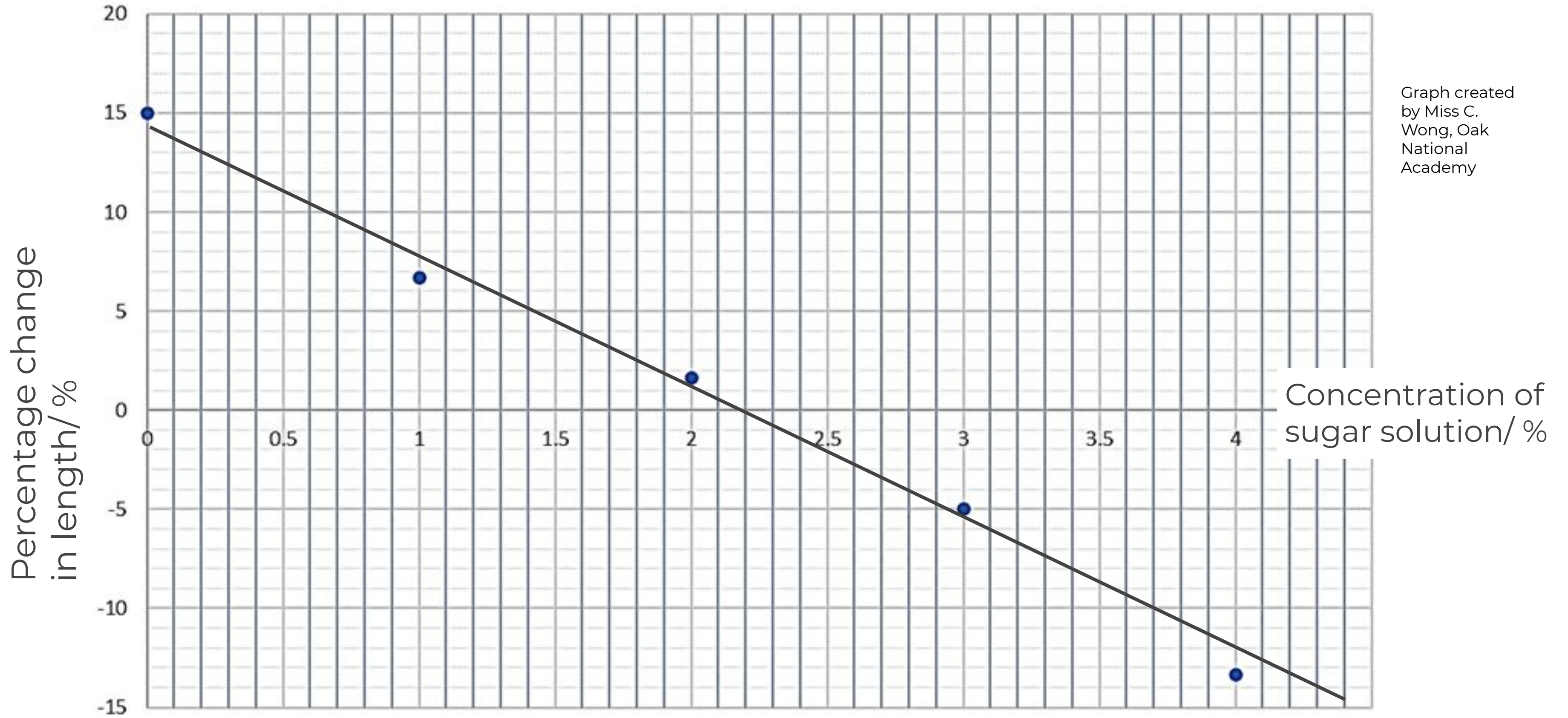
Concentration of sugar solution/%	Starting length/cm	Final length	Change in length	Percentage change in length/ %
4	6.00	5.20	-0.80	-13.33
3	6.00	5.70	-0.30	-5
2	6.00	6.10	0.10	1.67
1	6.00	6.40	0.40	6.67
0	6.00	6.90	0.90	15



Title: Percentage change in length of plant tissue in different concentrations of sugar solution



Title: Percentage change in length of plant tissue in different concentrations of sugar solution



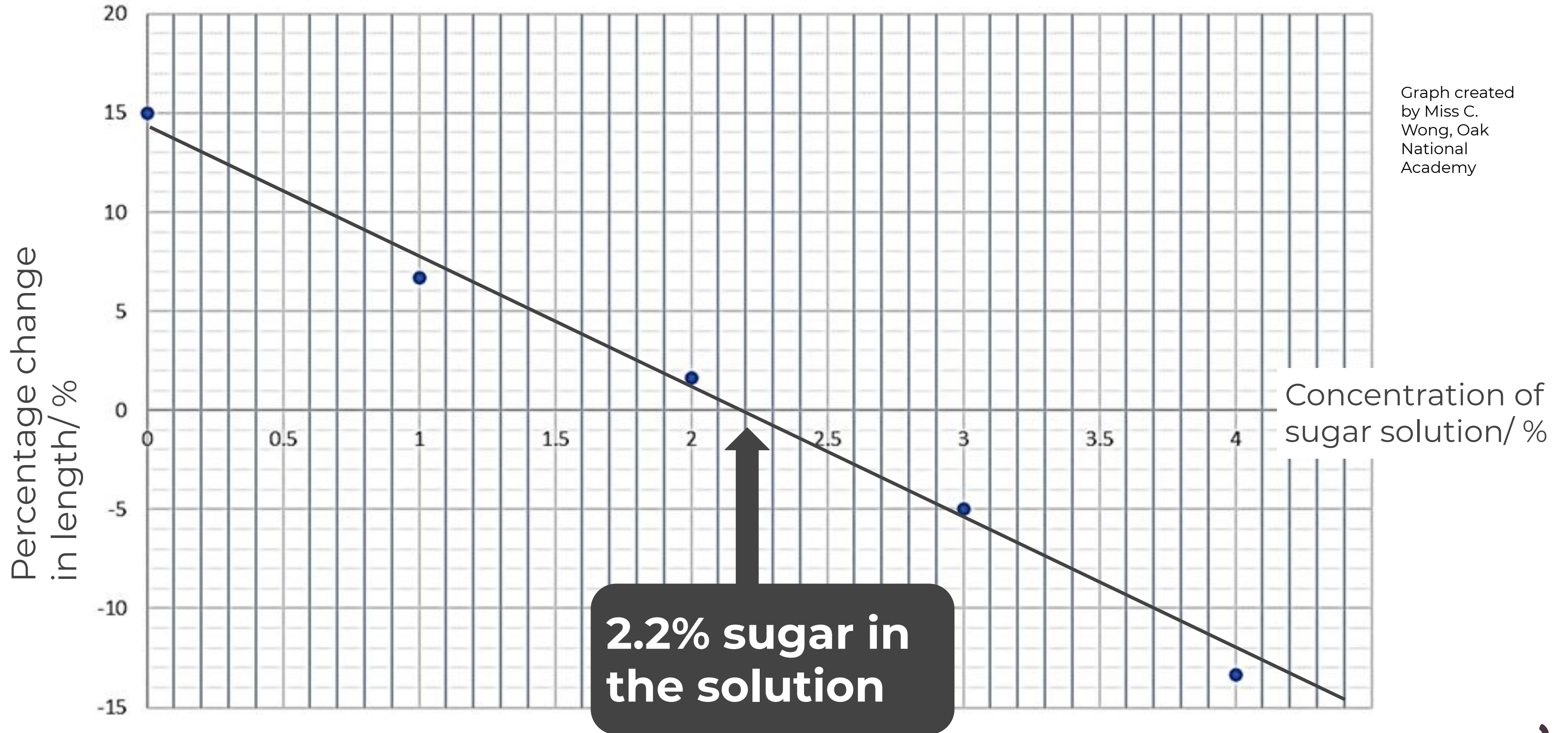
Graph created
by Miss C.
Wong, Oak
National
Academy

Concentration of
sugar solution / %

Percentage change
in length / %



Title: Percentage change in length of plant tissue in different concentrations of sugar solution



Pause the video to complete your task

Arrange the following sentences in the right order.

Read the x-intercept.

Find the change in mass or length Find the percentage change.

Draw the line of best fit.

Plot the data onto a graph.

Resume once you're finished



Independent practise



Independent practise

1. What is the formula to calculate percentage change in mass?
2. Why is percentage change used instead of change in mass?
3. Explain why some potato cylinders showed an increase in mass. For which range was this applicable?
4. Explain why some potato cylinders showed a decrease in mass. For which range was this applicable?
5. Estimate the concentration of sugar inside the potato cells. How can we tell that from the data?
6. How could we improve the estimate of the sugar solution in the cells?



Independent practise

Concentration of salt solution/ M	Starting mass/g	Final mass/g	Change in mass/g	Percentage change in mass/ %
1	6.1	5.2		
0.75	6.3		-0.5	-7.94
0.5	5.9	5.7	-0.2	
0.25		6.4	0.2	3.23
0	5.8	6.4		



Answers to independent practise

1. Percentage change = $\text{Change} \div \text{starting mass} \times 100$
2. To account for differences in starting mass.
3. Some potato cylinders gained mass because the surrounding solution was more dilute than in the cells. Water moved in by osmosis, so they gained water.
4. Some of the potato cylinders lost water by osmosis because the solution surrounding the cells was more concentrated. There was a net outward movement of water through the cell membrane and the cells shrunk, causing a loss of mass.
5. You need to read the point where the line crosses the horizontal axis (the x-intercept).
6. Repeat the experiment using a narrow range of concentration around the estimate from the graph e.g between 0.40 and 0.5M (0.40, 0.42, 0.44, 0.48M).



Answers to independent practise

Concentration of salt solution/ M	Starting mass/g	Final mass/g	Change in mass/g	Percentage change in mass/ %
1	6.1	5.2	-0.9	-14.75
0.75	6.3	5.8	-0.5	-7.94
0.5	5.9	5.7	-0.2	-3.39
0.25	6.2	6.4	0.2	3.23
0	5.8	6.4	0.6	10.34

